8-C-PRENYLFLAVONOIDS FROM THE SEED OF TEPHROSIA BRACTEOLATA

SAMI A. KHALID and PETER G. WATERMAN*

 $Phytochemistry, Research\,Laboratory, Department\,of\,Pharmaceutical\,Chemistry, University\,of\,Strathclyde, Glasgow, G1\,1XW, U.K.$

(Received 29 August 1980)

Key Word Index—*Tephrosia bracteolata*; Leguminosae; 8-C-prenyl-5,7-dioxygenated flavonoids; obovatin methyl ether; isopongaflavone; *trans*-tephrostachin; *trans*-anhydrotephrostachin.

Abstract—Four flavonoids in the seeds of *Tephrosia bracteolata* were identified as the flavanone obovatin methyl ether and the flavones isopongaflavone, *trans*-tephrostachin and *trans*-anhydrotephrostachin.

INTRODUCTION

Tephrosia bracteolata Guill. et Perr. is an erect undershrub common throughout savanna areas of tropical Africa [1]. As part of a study of the flavonoids of the Tephrosieae [2], we now report the results of an investigation of the seeds of this species.

RESULTS AND DISCUSSION

Extraction of the powdered seeds with petrol and then CHCl₃, followed in each case by column chromatography and preparative TLC over silica gel, gave four flavonoids. Two of these were identified on the basis of a comparison with published spectral data as obovatin methyl ether (1) and isopongaflavone (2). 1 had previously been isolated from *T. obovata* Merr. [3] and *T. praecana* Brummitt [4], and 2 from *Pongamia glabra* Vent. [5].

The major compound analysed for $C_{22}H_{22}O_5$. Both UV and IR spectra were typical of a flavone [6], and the latter indicated the presence of an OH substituent. The ¹HNMR spectrum permitted assignment of all twentytwo protons. Two multiplets centred at δ 7.55 (3H) and 7.90 (2H) could be attributed to an unsubstituted B-ring [2]. Four singlets, at δ 3.96 and 4.01 (3 H each) and 6.42 and 6.68 (1 H each), were indicative of a 5,7dimethoxyflavone nucleus substituted at either C-6 or C-8. The remaining nine protons were observed as an AB quartet (2 H, J = 17 Hz) centred at δ 6.75 and 6.90, a singlet at 1.90 (replaceable with D₂O), and a 6 H singlet at 1.49. These resonances are typical of the unusual 3hydroxy-3-methyl-trans-but-1-enyl side-chain [2, 7], and permit assignment of structure 3 or 4 to this flavonoid. The position of the C₅-substituent on the A-ring was resolved from a study of the resonance positions of the OMe carbons in the ¹³C NMR spectra. Both were observed at 56.3 ppm. Since in 3 the C-5 methoxyl would be forced, by steric hindrance, to exist outside the plane of the ring, it would resonate at least 2 ppm further downfield than this [8]. Structure 4 must therefore be assigned. 4 is the hitherto unrecorded trans-isomer of tephrostachin (5), reported from T. polystachyoides E. Mey [9], and the 5-methoxy derivative of lanceolatin-A (6) known from T. lanceolata Gamb [10], T. purpurea Pers.

The fourth flavonoid analysed for $C_{22}H_{20}O_4$. The UV spectrum was similar to that of 4 but the IR spectrum differed both in the absence of OH absorption and in exhibiting a weak band at 3100 cm⁻¹, typical of =: CH₂. The ¹H NMR spectrum again indicated a 5,7-dimethoxyflavone with C-8 substitution. The seven protons of the C-8 substituents once more included an AB quartet exhibiting trans-coupling together with broad

$$\begin{array}{c} R_3 \\ MeO \\ 7 \\ R_2 \\ R_1 \end{array} \begin{array}{c} 2 \\ 1 \\ 6 \\ 8 \end{array} \begin{array}{c} 3 \\ 5 \end{array}$$

	R_1	R ₂	R_3
3	OMe	CH:CHC(OH)Me ₂	Н
4	ОМе	н	t CH:CHC(OH)Me ₂
5	OMe	Н	CH:CHC(OH)Me ₂
6	Н	Н	t CH:CHC(OH)Me ₂
7	OMe	н	t CH:CHC(Me):CH ₂

^[11] and T. apollinea Link [2]. It seems most appropriate to assign to it the trivial name of trans-tephrostachin.

^{*}To whom all correspondence should be addressed.

singlets at δ 2.08 (3 H) and 5.12 (2 H) for Me and =CH₂ groups. These data are typical of the 3-methyl-trans-but-1,3-dienyl sidechain [7] and permit the assignment of structure 7 to this flavonoid. Compound 7, to which we have assigned the trivial name trans-anhydrote-phrostachin, appears to occur naturally rather than as an artefact from the dehydration of 4 [7].

A further blue fluorescent compound isolated from the MeOH extract of the seeds was identified as 6,7-dihydroxycoumarin (aesculetin) by direct comparison with an authentic sample. TLC analysis of this extract also indicated the occurrence of aesculin, the 6-O-glucoside of this coumarin.

EXPERIMENTAL

Plant material. Fruiting material of Tephrosia bracteolata Guill. et Perr. was collected in Kwabenya-Accra, Ghana, during December 1979. A voucher specimen, Enti FE-1959, has been deposited in the Herbarium of the Royal Botanic Gardens, Kew.

Extraction and separation. Powdered seeds (80g) were extracted separately with petrol (bp 60–80°), then CHCl₃, then 70% aq. MeOH. Concn of the petrol extract followed by column chromatography over Si gel gave, on elution with petrol–EtOAc (49:1), impure 2. Prep. TLC of this material (Si gel, solvent—petrol–EtOAc, 4:1) yielded pure 2 (56 mg). Further elution of the column with petrol–EtOAc (3:7) gave impure 7, subsequently purified by prep. TLC (Si gel, solvent—C₆H₆–EtOAc 3:2) to yield 46 mg. Concn of the CHCl₃ extract gave, on standing, 4 (211 mg). Column chromatography of the residue of extract over Si gel, eluting with petrol–EtOAc (4:1), with subsequent prep. TLC of the resulting eluate (Si gel, solvent—C₆H₆–EtOAc, 3:2) gave 1 (61 mg). Concn of the MeOH extract and prep. TLC of the EtOAc–soluble fraction (Si gel, solvent—CHCl₃–MeOH, 17:1) gave 6,7-dihydroxycoumarin (27 mg).

Obovatin methyl ether (1). Needles from petrol-EtOAc, mp 135° (lit. [4] 125-127°). Found: M^+ 336.1369; $C_{21}H_{20}O_4$ requires: 336.1361. UV, IR and ¹H NMR as lit. [3]. MS m/z (rel. int.): 336 (36), 321 (38), 218 (12), 217 (100).

Isopongaflavone (2). Plates from CHCl₃-MeOH, mp 201–205° (lit. [5] 215–216°). Found: M⁺ 334.1216; C₂₁H₁₈O₄ requires: 334.1205. UV and IR as lit. [5]. ¹H NMR (CDCl₃): δ 1.49 (6 H, s, 2"-Me₂), 3.94 (3 H, s, 5-OMe), 5.66, 6.84 (2 H, ABq, J = 10 Hz, 3" and 4"-H), 6.33 (1 H, s, 6-H), 6.67 (1 H, s, 3-H), 7.45–7.55 (3 H, m, 2',4',6'-H), 7.80–7.95 (2 H, m, 3',5'-H). MS m/z (rel. int.): 334 (48), 320 (16), 217 (23).

trans-Tephrostachin (4). Needles from CHCl₃, mp 215–220°. Found: M⁺ 366.1469; C₂₂H₂₂O₅ requires: 366.1467. UV λ_{max} nm: 265, 345. IR ν_{max} cm⁻¹: 3350, 1640. ¹H NMR (CDCl₃): δ 1.49 (6 H, s, 3"-Me₂), 1.90 (1 H, s, replaceable by D₂O, 3"-OH), 3.96, 4.01 (2 × 3 H, 2 × s, 5 and 7-OMe), 6.42 (1 H, s, 6-H), 6.68 (1 H, s, 3-H), 6.75, 6.90 (2 H, ABq, J = 17 Hz, 2" and 1"-H),

7.47–7.62 (3 H, m, 2',4'6'-H), 7.84–7.98 (2 H, m, 3',5'-H).

13°C NMR (DMSO-d₆): ppm 30.2 (3"-Me₂), 56.3 (5 and 7-OMe),
70.4 (C-3"), 93.1 (C-6), 107.3, 108.0 (C-4a, C-8), 108.5 (C-3), 114.7 (C-1"), 126.7, 126.8 (C-2', C-6'), 129.7 (C-3', C-5'), 132.1 (C-4', C-1'), 143.7 (C-2"), 155.9 (C-8a), 160.0, 161.1 (C-5, C-2), 162.2 (C-7),
177.6 (C-4). MS m/z (rel. int.): 366 (20), 351 (11), 349 (38), 348 (100), 320 (21), 319 (81), 317 (27), 287 (31), 225 (13).

trans-Anhydrotephrostachin (7). Clusters from EtOAc, mp 128–134°. Found: M⁺ 348.1364; $C_{22}H_{20}O_4$ requires 348.1361. UV λ_{max} nm: 264, 277, 350. IR v_{max} cm⁻¹: 3100, 1640. ¹H NMR (CDCl₃): δ 2.08 (3 H, s, 3"-Me), 3.92 (6 H, s, 5 and 7-OMe), 5.12 (2 H, s, 3"-=CH₂), 6.49 (1 H, s, 6-H), 6.62 (1 H, s, 3-H), 6.89, 7.31 (2 H, ABq, J = 17 Hz, 2" and 1"-H), 7.47–7.60 (3 H, m, 2',4',6'-H), 7.85–8.00 (2 H, m, 3',5'-H). MS m/z (rel. int.): 348 (69), 319 (16), 310 (27), 309 (15), 295 (27), 230 (16), 125 (15), 95 (100).

Acknowledgements—The authors thank Mr. A. A. Enti, Forestry Enterprises (Ghana), for arranging the supply of plant material and Dr. P. Bladon, Department of Chemistry, University of Strathclyde, for ¹³C NMR spectra. One of us (S.A.K.) wishes to thank the University of Khartoum for the award of a scholarship. The ¹H NMR spectrometer used in this study was purchased on Grant No. A, 80754 from the Science Research Council. Dr. R. M. Polhill, Royal Botanic Gardens, Kew, kindly confirmed the identity of the plant material.

REFERENCES

- Hutchinson, J. and Dalziel, J. M. (1958) Flora of West Tropical Africa, 2nd edn. Vol. 1(2), p. 530. Crown Agents, London.
- Waterman, P. G. and Khalid, S. A. (1980) Phytochemistry 19, 909.
- Chen, Y.-L., Wang, Y.-S., Lin, Y.-L., Munakata, K. and Ohta, K. (1978) Agric. Biol. Chem. 42, 2431.
- 4. Camele, G., Delle Monache, F., Delle Monache, G. and Marini Bettolo, G. B. (1980) *Phytochemistry* 19, 707.
- Roy, D., Sharma, N. N. and Khanna, R. N. (1977) Indian J. Chem. 15B, 1138.
- Harborne, J. B., Mabry, T. J. and Mabry, H. (1975) The Flavonoids. Chapman & Hall, London.
- Gray, A. I., Waigh, R. D. and Waterman, P. G. (1975) J. Chem. Soc. Perkin Trans. 1, 488.
- Panichpol, K. and Waterman, P. G. (1978) Phytochemistry 17, 1363.
- Vleggar, R., Smalberger, T. M. and de Waal, H. L. (1973) J. S. Afr. Chem. Inst. 26, 71.
- Ayengar, K. N. M., Rama Sastry, B. V. and Rangaswami, S. (1973) Indian J. Chem. 11B, 85.
- Venkata Rao, E. and Ranga Raju, N. (1979) Phytochemistry 18, 1581.